

# **Safety Data Sheet**

Review Date : 4-Oct-2023

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## **Section 1 - Chemical Product and Company Identification**

Product Name	Sodium Hydroxide 0.2N Solution			
Product Code	27382			
CAS No	1310-73-2			
Use for	Laboratory Chemicals.			
Company Name	Sisco Research Laboratories Pvt. Ltd.			
Address	608, B Wing, Satellite Gazebo, Andheri Ghatkopar Link Road, Andheri (E), Mumbai - 400 099, India			

#### Section 2 - Composition/Information on Ingredients

CAS#,	Chemical Name,	Percent,	EC Number
1310-73-2	Sodium hydroxide	<=100%	215-185-5

## Section 3 - Hazards Identification

**Danger!** Causes eye and skin burns. Causes digestive and respiratory tract burns. Hygroscopic (absorbs moisture from the air).

Target Organs: Eyes, skin, mucous membranes.

#### Potential Health Effects

Eye: Causes eye burns. May cause blindness. May cause chemical conjunctivitis and corneal damage.

Skin: Causes skin burns. May cause deep, penetrating ulcers of the skin.

**Ingestion:** May cause severe and permanent damage to the digestive tract. Causes gastrointestinal tract burns. May cause perforation of the digestive tract. Causes severe pain, nausea, vomiting, diarrhea, and shock.

**Inhalation:** Irritation may lead to chemical pneumonitis and pulmonary edema. Causes severe irritation of upper respiratory tract with coughing, burns, breathing difficulty, and possible coma. Causes chemical burns to the respiratory tract.

Chronic: Prolonged or repeated skin contact may cause dermatitis. Effects may be delayed.

### Section 4 - First Aid Measures

**Eyes:** In case of contact, immediately flush eyes with plenty of water for a t least 15 minutes. Get medical aid immediately.

**Skin:** In case of contact, immediately flush skin with plenty of water for at least 15 minutes while removing contaminated clothing and shoes. Get medical aid immediately. Wash clothing before reuse.

**Ingestion:** If swallowed, do NOT induce vomiting. Get medical aid immediately. If victim is fully conscious, give a cupful of water. Never give anything by mouth to an unconscious person.

**Inhalation:** If inhaled, remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical aid.

Notes to Physician: Treat symptomatically and supportively.

### Section 5 - Fire Fighting Measures



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**General Information:** As in any fire, wear a self-contained breathing apparatus in pressure-demand, MSHA/NIOSH (approved or equivalent), and full protective gear. Use water spray to keep fire-exposed containers cool. Use water with caution and in flooding amounts. Contact with moisture or water may generate sufficient heat to ignite nearby combustible materials. Contact with metals may evolve flammable hydrogen gas.

**Extinguishing Media:** Substance is noncombustible; use agent most appropriate to extinguish surrounding fire. Do NOT get water inside containers.

Flash Point: Not applicable.

Autoignition Temperature: Not applicable.

Explosion Limits, Lower: Not available.

Upper: Not available.

NFPA Rating: (estimated) Health: 3; Flammability: 0; Instability: 1

### Section 6 - Accidental Release Measures

**General Information:** Use proper personal protective equipment as indicated in Section 8. **Spills/Leaks:** Vacuum or sweep up material and place into a suitable disposal container. Avoid runoff into storm sewers and ditches which lead to waterways. Clean up spills immediately, observing precautions in the Protective Equipment section. Avoid generating dusty conditions. Provide ventilation. Do not get water on spilled substances or inside containers

## Section 7 - Handling and Storage

Handling: Wash thoroughly after handling. Do not allow water to get into the container because of violent reaction.
Minimize dust generation and accumulation. Do not get in eyes, on skin, or on clothing. Keep container tightly closed.
Avoid ingestion and inhalation. Discard contaminated shoes. Use only with adequate ventilation.
Storage: Store in a tightly closed container. Store in a cool, dry, well-ventilated area away from incompatible substances. Keep away from metals. Corrosives area. Keep away from acids. Store protected from moisture.
Containers must be tightly closed to prevent the conversion of NaOH to sodium carbonate by the CO2 in air.

## Section 8 - Exposure Control / Personal Protection

## **Personal Protective Equipment**

<b>Respiratory Protection</b>	: Where risk assessment shows air-purifying respirators are appropriate use a			
	full-face particle respirator type N99 (US) or type P2 (EN 143) respirator			
	cartridges as a backup to engineering controls. If the respirator is the sole			
	means of protection, use a full-face supplied air respirator. Use respirators			
	and components tested and approved under appropriate government standards			
	such as NIOSH (US) or CEN (EU).			
Hand Protection:	The selected protective gloves have to satisfy the specifications of EU Directive			
	89/686/EEC and the standard EN 374 derived from it. Handle with gloves.			
Eye Protection:	Safety glasses			
Skin and body protection	<b>on:</b> Choose body protection according to the amount and concentration of the dangerous substance at the work			
	place.			
	*			
Hygiene measures:	Avoid contact with skin, eyes and clothing. Wash hands before breaks and			
immediately after handling the product.				

## **Section 9 - Physical and Chemical Properties**



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Physical State: Solid Appearance: white Melting Point: 318 deg C Molecular Formula:NaOH Molecular Weight: 40.00

Section 10 - Stability and Reactivity

Chemical Stability: Stable at room temperature in closed containers under normal storage and handling conditions. Conditions to Avoid: Moisture, contact with water, exposure to moist air or water, prolonged exposure to air. Incompatibilities with Other Materials: Water, metals, acids, aluminum, zinc, tin, nitromethane, leather, flammable liquids, organic halogens, wool.

Hazardous Decomposition Products: Toxic fumes of sodium oxide.

Hazardous Polymerization: Will not occur.

## **Section 11 - Toxicological Information**

Acute toxicity:	Inhalation, mouse: $LC50 = 1200 \text{ mg/m3/2H}$ ;		
	Inhalation, rat: $LC50 = 2300 \text{ mg/m3/2H}$ ;		
Oral, mouse: $LD50 = 6600 \text{ mg/kg}$ ;			
Oral, mouse: $LD50 = 6600 \text{ mg/kg};$			
Oral, rat: LD50 = 4090 mg/kg;			
Irritation and corrosion: No data available			
Sensitisation:	No data available		
Chronic exposu	re: IARC: No component of this product present at levels greater than or equal to 0.1% is identified as probable, possible or confirmed human carcinogen by IARC.		
Signs And Sym	ntoms		

Signs And Symptoms **Of Exposure:** No data available

## **Route Of Exposure**

Inhalation: No data available Skin : No data available

No data available Eyes:

**Ingestion:** No data available

## **Section 12 - Ecological Information**

No information available.

Section 13 - Disposal Considerations

**Product:** Observe all federal, state, and local environmental regulations. Contact a licensed professional waste disposal service to dispose of this material. Dissolve or mix the material with a combustible solvent and burn in a chemical incinerator equipped with an afterburner and scrubber. Dispose of as unused product.

**Contaminated packaging:** 

**Section 14 - Transport Information** 



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Product Code	27382			
	IATA	4	IMO	<b>RID/ADR</b>
Shipping Name:	SODIUM HYDROXIDE, SOLID			
Hazard Class:	8	8	8	
UN Number:	1823	1823	3	1823
<b>Packing Group:</b>	II	II	II	

**Section 15 - Regulatory Information** 

This safety datasheet complies with the requirements of Regulation (EC) No. 1907/2006.

## **Section 16 - Other Information**

Sisco Research Laboratories Pvt. Ltd. provides the information contained herein in good faith but makes no representation as to its comprehensiveness or accuracy. This document is intended only as a guide to the appropriate precautionary handling of the material by a properly trained person using this product. Individuals receiving the information must exercise their independent judgment in determining its appropriateness for a particular purpose.